

# A Novel CA4P Polymeric Nanoparticle for Murine Hepatoma Therapy

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## Electronic Supplementary Information

**Abstract** Combretastatin A4 phosphate (CA4P) is a potent vascular disrupting agent with good water solubility. However, it is only effective at high doses, which decreases clinical applicability. Herein, we designed stable CA4P polymeric nanoparticles (CA4P NPs) consisting of various cholesterol derivatives, and with a drug loading efficacy of 93%. The nanoparticles released CA4P in a sustained manner and achieved a 72% inhibition rate in the murine H22 liver tumor model, which was about 2.9-fold higher than that of free CA4P (24.6%). Furthermore, the carrier components of CA4P NPs were metabolized to arginine, cholesterol, ethanol and poly(ethylene glycol) *in vivo*; therefore, the CA4P NPs are safe and have significant potential for clinical translation.

**Keywords** CA4P polymeric nanoparticle; Cholesterol derivatives; Vascular disrupting agent; Phosphate-guanidine coordination; Drug controlled release

**Citation:** Liu, Z. L.; Ren, X. T.; Huang, Y.; Sun, J. L.; Wang, X. S.; Zheng, M. F.; Cui, L. J.; Zhang, X. F.; Tang, Z. H. A novel CA4P polymeric nanoparticle for murine hepatoma therapy. *Chinese J. Polym. Sci.* 2023, 41, 1223–1229.

## INTRODUCTION

The tumor vascular system provides oxygen and nutrients to the rapidly proliferating cells and is therefore critical for tumor growth and metastasis.<sup>[1–3]</sup> Thus, vascular disrupting agents (VDAs) are promising anti-tumor drugs since they can “starve” the tumor and induce necrosis by blocking the supply of oxygen and nutrients.<sup>[4–6]</sup> Since VDAs inhibit tumor growth through “remote” action rather than direct contact,<sup>[7–10]</sup> they can overcome the limitations of traditional chemotherapy drugs that must penetrate deep into the tumor to be effective.<sup>[11,12]</sup>

Combretastatin A4 (CA4) is a potent VDA with broad-spectrum anti-tumor activity<sup>[13–15]</sup> that inhibits tumor vascular endothelial cells by interfering with the nucleation process of microtubulin.<sup>[16–19]</sup> A phosphate derivative of CA4 (CA4P) was designed to improve its aqueous solubility.<sup>[20]</sup> CA4P is reduced to CA4 prodrug in the tumor microenvironment with high levels of alkaline phosphatase content. Although CA4P has been tested in a phase II-III clinical trial in combination with a partial drug regimen,<sup>[21]</sup> it is effective only at high

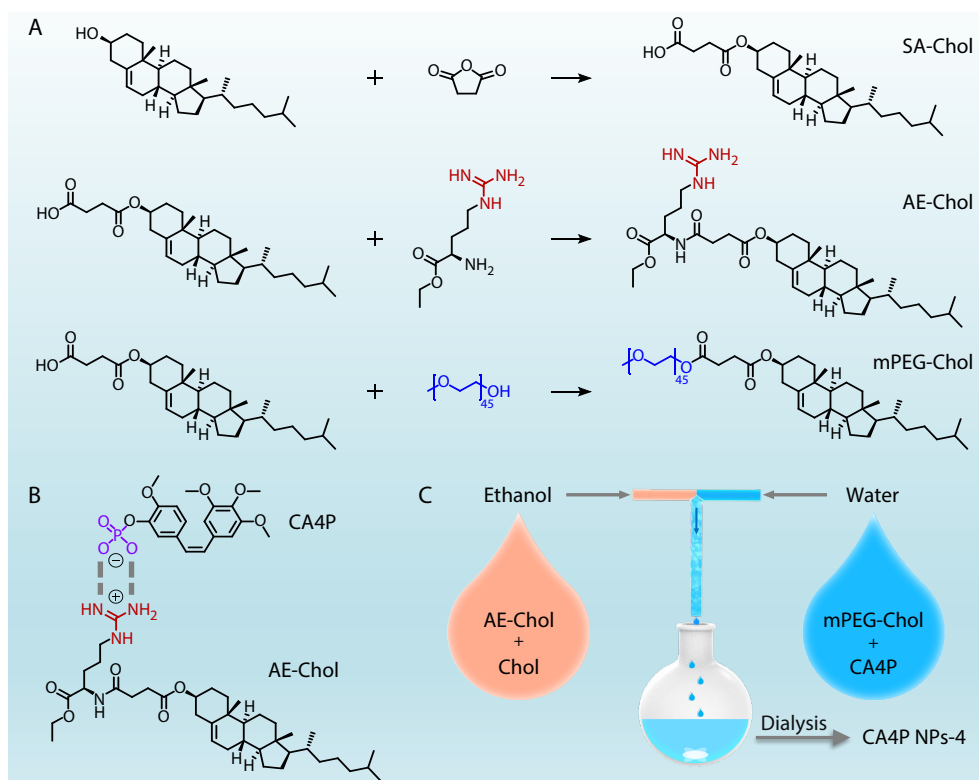
doses (~100 mg/kg) that can cause severe cardiotoxicity.<sup>[22–24]</sup> Thus, clinical studies on CA4P have been considerably delayed. We previously found that CA4 loaded into nanocarriers can be delivered selectively to the tumor vessels,<sup>[25]</sup> which can destruct tumor vascular and cause persistent tumor necrosis, eventually resulting in more effective growth suppression.<sup>[26]</sup> However, we had synthesized CA4 polymeric nanoparticles by attaching the drug to poly(L-glutamic acid)-graft-poly(ethylene glycol) *via* esterification of carboxyl and phenolic hydroxyl, which may cause systemic toxicity if CA4 is accidentally released in the bloodstream.<sup>[27]</sup> Given that CA4P is less toxic compared to CA4,<sup>[28–30]</sup> we surmised that CA4P nanodrugs would be more effective and have a greater potential for clinical translation.

Herein, we designed a novel polymeric nanoparticle delivery system by synthesizing and assembling various cholesterol derivatives into CA4P nanoparticles (CA4P NPs) through the coordination of the phosphate group in CA4P with the guanidine group in guanidine-based cholesterol (AE-Chol) (Fig. 1). Stable CA4P NPs with 93% drug loading efficacy (CA4P NPs-4) were obtained after screening various loading methods and achieved sustained release of CA4P. The CA4P NPs-4 inhibited H22 liver tumor growth in a mouse model by 72%, which was about 2.9-fold higher than that of free CA4P (24.6%), and significantly increased necrosis of tumor cells.

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Received November 19, 2022; Accepted December 6, 2022; Published online February 10, 2023



**Fig. 1** Synthesis of various cholesterol derivatives and the preparation of CA4P NPs-4.

The CA4P NPs-4 were metabolized *in vivo* to arginine, cholesterol, ethanol and poly(ethylene glycol). Therefore, CA4P NPs-4 have high biological safety and show potential for clinical translation.

## EXPERIMENTAL

### Reagents

Cholesterol (Chol), H-Arg-OEt-2HCl (AE), succinic anhydride (SA) and dimethylaminopyridine (DMAP) were purchased from Shanghai Macklin Biochemical Co., Ltd. *N,N*-carbonyldiimidazole (CDI) and *N,N*-diisopropylethylamine (DIPEA) were purchased from Sun Chemical Technology Co., Ltd. Methoxypoly(ethylene glycol) (mPEG-OH,  $M_n=2.0 \times 10^3$  g/mol) was purchased from Jenkem Technology Co., Ltd. Combretastatin A4 phosphate (CA4P) was purchased from Shanghai Bide Pharmaceutical Technology Co., Ltd.

### Instruments

$^1\text{H-NMR}$  spectra were recorded on a Bruker AV 300 NMR (China) spectrometer in chloroform-*d*. Electron spray ionization mass spectrometry (ESI-MS) and matrix-assisted laser desorption/ionization mass spectrometry time of flight mass spectrometry (MALDI-TOF MS) were performed using a Thermo LTQ XL ion trap mass spectrometer. The zeta potentials and particle sizes were measured using Malvern Zetasizer Pro (United Kingdom). The ultraviolet/visible absorption (UV/Vis) spectra were measured using a UV-2401PC spectrophotometer (Japan). UV high-performance liquid chromatography (UV-HPLC) was performed using the Dalian Elite Analytical Instrument (China) and INano L microfluidic device

(Micro&Nano Biologics Co., Ltd., China).

### Cell Lines and Animals

The murine H22 hepatoma cell line was purchased from Shanghai Bogoo Biotechnology Co., Ltd., (China). Six to eight-weeks old female BALB/c mice (16–18 g) were purchased from Beijing Vital River Laboratory Animal Technology Co., Ltd., (China). All animal experiments were performed according to the Guidelines for Care and Use of Laboratory Animals of Jilin University, and approved by the Animal Ethics Committee of Jilin University.

### Preparation of Cholesterol Derivatives

#### SA-Chol

Chol (1.0 equiv.), SA (1.5 equiv.) and DMAP (0.2 equiv.) were added to a round-bottomed flask, dissolved with anhydrous pyridine and reacted at 80 °C for 3 h. The pyridine was drained and then dissolved in anhydrous dichloromethane (DCM), and washed with 0.1 mol/L HCl. After recrystallization in *n*-hexane, the white solid SA-Chol was obtained (yield 95.5%) and characterized by  $^1\text{H-NMR}$  and ESI-MS (ESI<sup>+</sup>).

#### AE-Chol

SA-Chol (1.0 equiv.) and CDI (2.0 equiv.) were added to a round-bottomed flask and dissolved in anhydrous DCM, reacted overnight at 30 °C, and washed with brine to obtain Chol-CDI. Arg-EED-2HCl (1.0 equiv.) and DIPEA (1.0 equiv.) were dissolved in anhydrous *N,N*-dimethylformamide (DMF), and Chol-CDI (1.2 equiv.) was dissolved in anhydrous DCM and added to the mixture. The reaction was performed overnight at 45 °C. After precipitation and washing with ethyl ether for 3 times, the white solid AE-Chol was obtained (yield 63.2%) and characterized by

$^1\text{H-NMR}$  and ESI-MS ( $\text{ESI}^+$ ).

#### mPEG-Chol

mPEG-OH (1.0 equiv.) was dissolved in anhydrous DMF and SA-Chol (1.5 equiv.), and DIC (1.5 equiv.) and DMAP (1.0 equiv.) were dissolved in anhydrous DCM. Both solutions were mixed and reacted at 35 °C for 48 h. The mixture was precipitated by ethyl ether, and the precipitate was redissolved in DMF and dialyzed against water for 48 h (MWCO=1.0 kDa). The white solid mPEG-Chol was obtained (yield 93.5%) after freeze-drying and characterized by  $^1\text{H-NMR}$  and MALDI-TOF MS.

#### CA4P Loading by Film Hydration and Microfluidic

##### Film hydration

AE-Chol (1.0 equiv.), mPEG-Chol (1.0 equiv.) and Chol (0 or 1.0 equiv.) were added to a round-bottomed flask and dissolved in DCM. The solvent was removed by rotary evaporation, leaving a residual film. CA4P (1.0 equiv.) was dissolved in water and heated to 50 °C, and then added to the round-bottomed flask containing the carrier film and incubated at 50 °C for 1 h to obtain CA4P NPs-1 and CA4P NPs-3.

##### Microfluidics

AE-Chol and Chol were dissolved in ethanol, and mPEG-Chol and CA4P were dissolved in water. Both solutions were mixed by INano L and ultrafiltered by Millipore Amicon Ultra (MWCO=3.0 kDa) to obtain CA4P NPs-2 and CA4P NPs-4.

The drug loading efficiency (DLE) and the drug loading content (DLC) of the nanoparticles above are calculated ac-

ording to the following formulas: DLE = the weight of loaded CA4P/the weight of feeding CA4P  $\times$  100%; DLC = the weight of loaded CA4P/the weight of CA4P nanoparticles  $\times$  100%.

#### Establishment of *In vivo* Tumor Model

The murine H22 tumor model was established by injecting BALB/c mice with H22 cells ( $3 \times 10^5$  cells per mouse) into their right abdomen. When the tumors reached approximately 100 mm<sup>3</sup>, the mice were randomly divided into 3 groups ( $n=5$ ) and respectively injected with PBS, CA4P (30 mg/kg) and CA4P NPs (30 mg/kg CA4P). The tumor sizes and body weight changes were recorded every other day. The tumor volume was calculated as follows: Tumor volume = 0.5  $\times$  length  $\times$  width<sup>2</sup>. The mice were sacrificed at the end of the treatment. The normal tissues and tumors were excised for routine H&E staining.

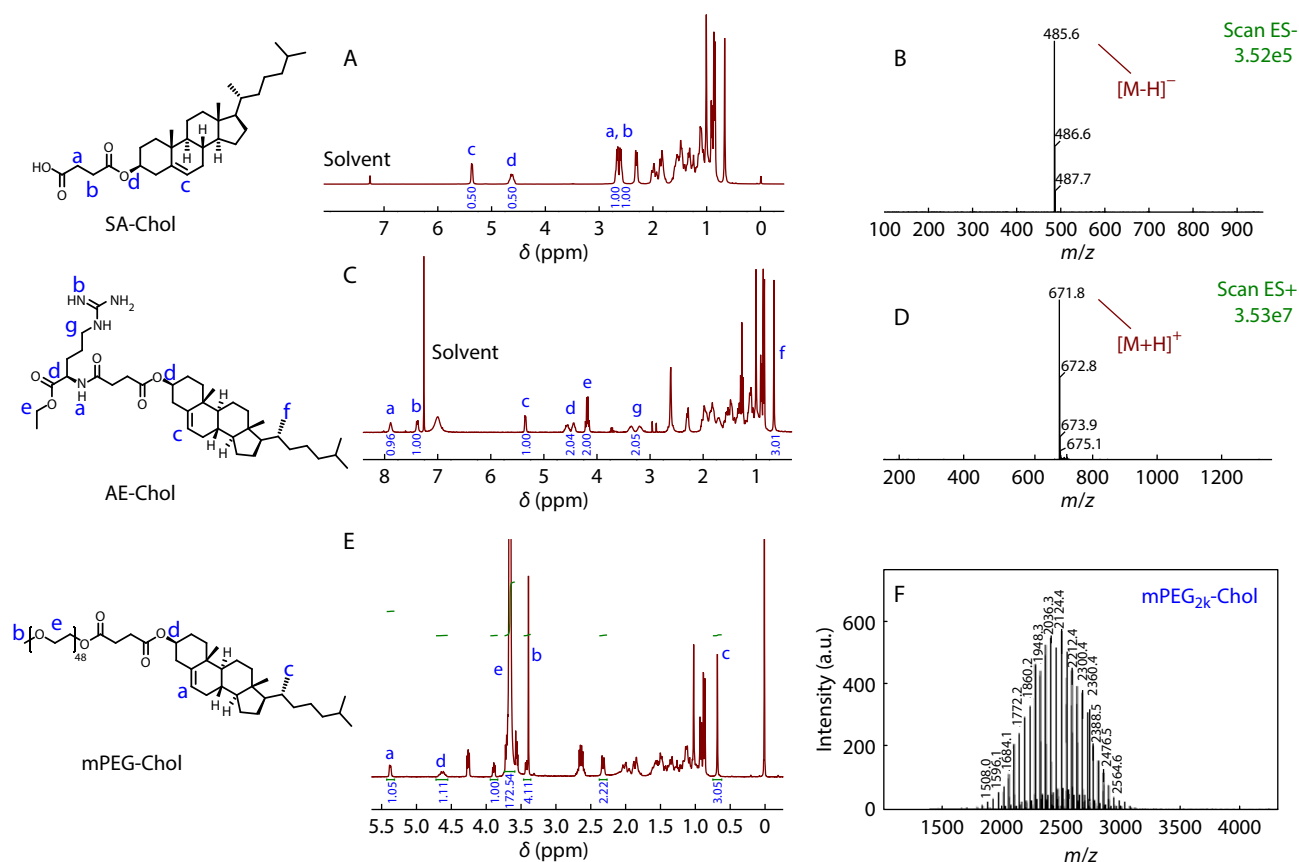
#### Statistical Analysis

Data are presented as the mean  $\pm$  standard deviation ( $n=3$ ) and compared by the Student's *t*-test. Statistical significances were set as follows: \* $p < 0.05$ ; highly significant: \*\* $p < 0.01$ ; ns: not significant.

## RESULTS AND DISCUSSION

#### Characterization of SA-Chol, AE-Chol and mPEG-Chol

The  $^1\text{H-NMR}$  spectrum of SA-Chol (Fig. 2A) shows peaks a and b of the methylene group at  $\delta=2.61$  ppm and  $\delta=2.65$  ppm, respectively, which are formed by the ring opening of



**Fig. 2** The  $^1\text{H-NMR}$  and ESI MS spectra of (A, B) SA-Chol, (C, D) AE-Chol, and the (E)  $^1\text{H-NMR}$  and (F) MALDI-TOF MS spectra of mPEG-Chol.

cholesterol-triggered succinic anhydride. The peaks c and d of the methyne group at  $\delta=5.37$  ppm and  $\delta=4.64$  ppm corresponded to cholesterol. The ratio of the peak a, peak b, peak c and peak d areas was 0.5:0.5:1:1, which indicated successful synthesis of SA-Chol. The ESI MS ( $m/z$ ) for  $C_{31}H_{50}O_4$   $[M-H]^-$  was 485.6 (Fig. 2B). In the  $^1H$ -NMR spectrum of AE-Chol (Fig. 2C), the ratio of peak e ( $\delta=4.17$  ppm) to peak f ( $\delta=0.67$  ppm) area was 2:3, which indicated the successful acquisition of AE-Chol. The ESI MS ( $m/z$ ) for  $C_{39}H_{66}N_4O_5$   $[M+H]^+$  was 671.8 (Fig. 2D). In the  $^1H$ -NMR of mPEG-Chol (Fig. 2E), the ratio of peak e ( $\delta=3.64$  ppm,  $-CH_2-CH_2-$  of mPEG) to peak a ( $\delta=5.37$  ppm,  $-CH-$  of cholesterol) area was 172.5:1.0, which was indicative of the successful condensation of mPEG-OH with Chol into mPEG-Chol. As shown in Fig. 2(F), the MALDI-TOF MS ( $m/z$ ) for mPEG-Chol was 2594.2, which is 469.8 (the molecular weight of the bonded cholesterol group) more than that of mPEG-OH (2124.4) and confirmed the successful synthesis of mPEG-Chol (Fig. S1 in the electronic supplementary information, ESI).

### Characterization of CA4P NPs

As the guanidine<sup>+</sup>/phosphate<sup>-</sup> salt bridge bound tightly in neutral and basic solution, four CA4P NPs were obtained through the salt-bridging interaction of the phosphate in CA4P with the guanidine group in AE-Chol by film hydration and microfluidic methods. The DLE by film hydration (CA4P NPs-1, Fig. 3A) and microfluidic (CA4P NPs-2, Fig. 3B) without the addition of Chol were 21.63% and 18.43%, the zeta potentials for both NPs were positive, and the particle size distributions were not uniform. Although the guanidine group coordinated with phosphate to form a hydrophobic core together with the hydrophobic Chol in mPEG-Chol, the results showed that the encapsulation of CA4P was ineffective, and the particle structure was not stable enough. Thus, Chol was added to form more

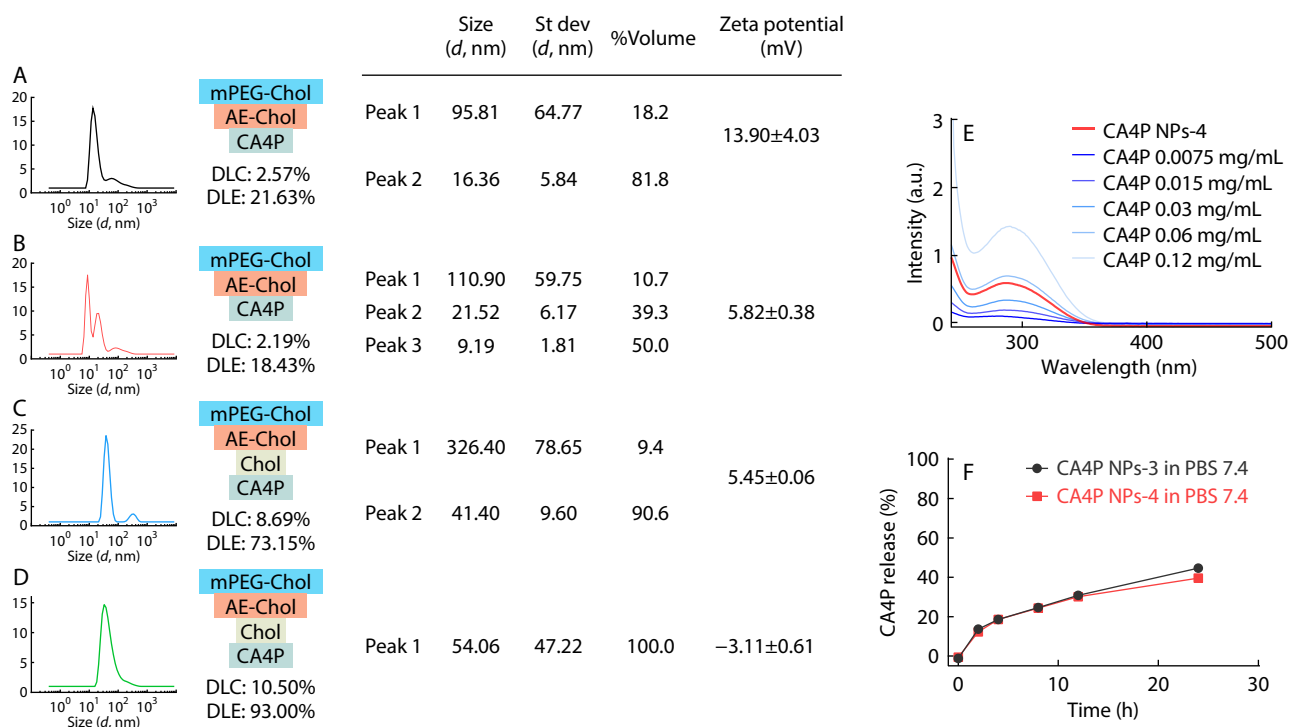
stable nanoparticles. After addition of Chol, the DLEs increased with both film hydration (CA4P NPs-3, DLE=73.15%) and microfluidic (CA4P NPs-4, DLE=93.00%) methods (Fig. 3C). However, the zeta potential of CA4P NPs-3 was still positive and the particle size was not homogeneous. In contrast, the particle size of CA4P NPs-4 (Fig. 3D) obtained by microfluidic method was homogeneous, which was about 54.1 nm, and showed good stability in PBS buffer (Fig. S2 in ESI). Thus, the microfluidic method was the most effective way for synthesizing CA4P NPs, and the content of each component was optimized accordingly (Fig. S2 in ESI). After quantifying the CA4P absorption peak (wavelength=290 nm) by UV/Vis, AE-Chol, Chol, mPEG-Chol and CA4P added in the proportion of 1:0.5:1:1 resulted in 93% DLE and 10.5% DLC (Fig. 3E). Meanwhile, as we can see in Fig. 3(F), the CA4P NPs-4 achieved a sustained release of CA4P (12.2% at 2 h and 39.6% at 24 h), therefore, CA4P NPs-4 was used for the subsequent experiments.

### In vivo Anti-tumor Efficacy

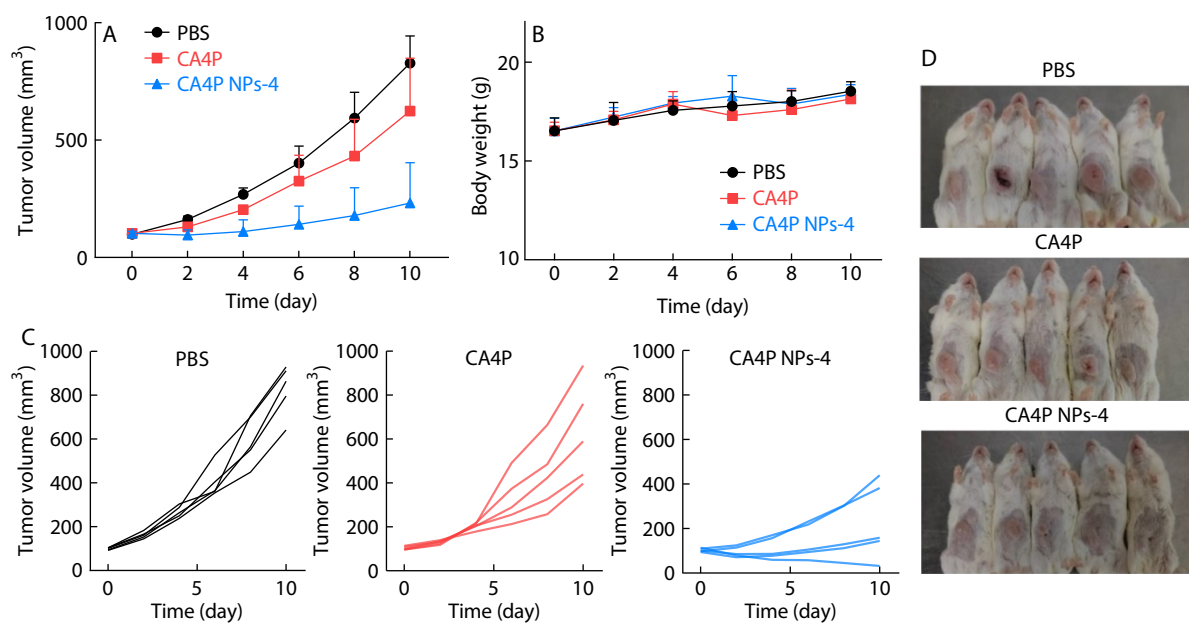
The mice in the PBS control group show rapid tumor growth, while both CA4P (30 mg/kg) and CA4P NPs-4 (30 mg/kg CA4P) inhibited the growth of H22 tumors in BALB/c mice (Figs. 4A and 4C). In fact, one mouse in the CA4P NPs-4 group develops tumor crusting and shedding (Fig. 4D) without causing any obvious change in body weight (Fig. 4B). While both CA4P and CA4P NPs-4 were safe at the dose of 30 mg/kg, CA4P NPs-4 inhibited tumor growth by 72.04% compared to the 24.64% inhibition rate achieved by CA4P.

### Immunohistochemical Staining

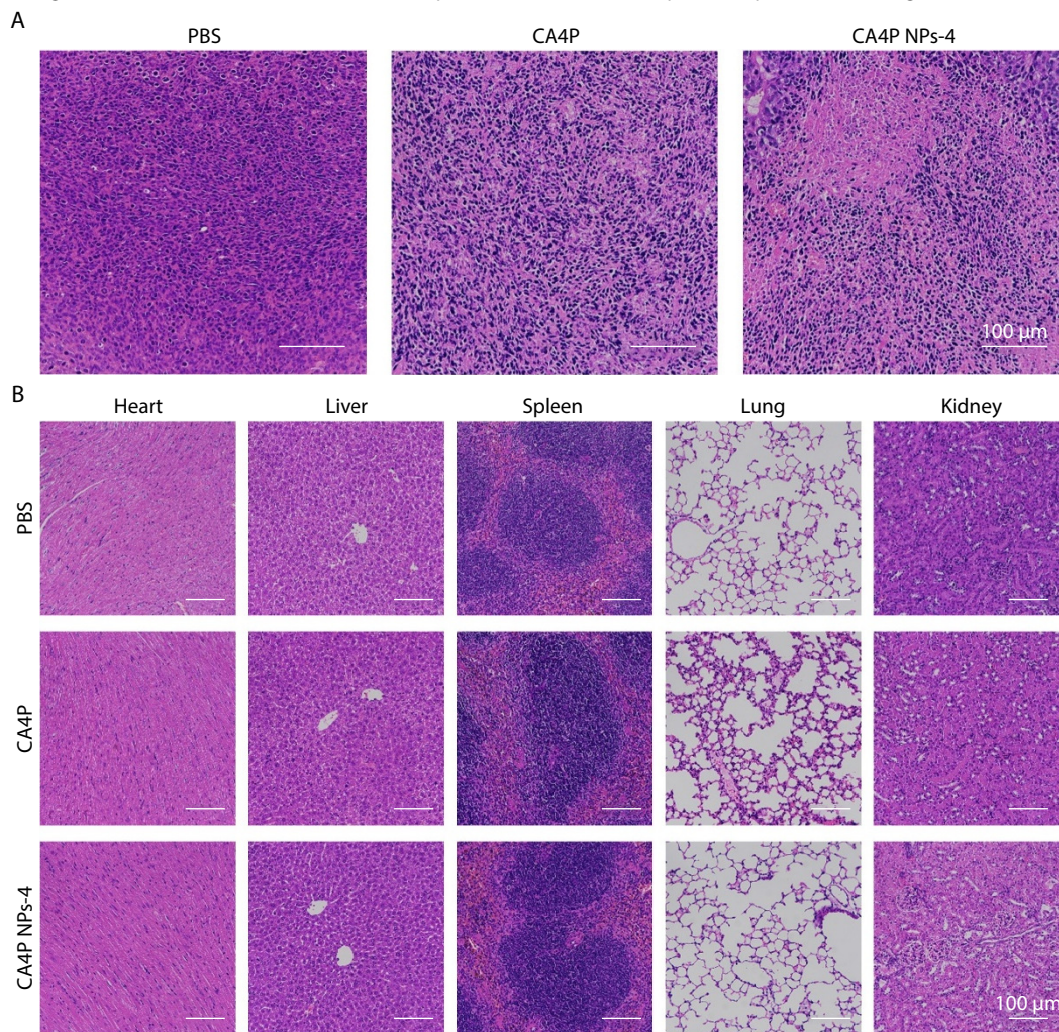
To further assess the anti-tumor effects of the CA4P NPs, we analyzed the histopathological changes in the tumors and normal tissues. As shown in Fig. 5(A), the tumors in the PBS-



**Fig. 3** The characterization of (A) CA4P NPs-1, (B) CA4P NPs-2, (C) CA4P NPs-3 and (D) CA4P NPs-4 via film hydration or microfluidic; (E) The DLC and DLE measurements of CA4P NPs-4; (F) The CA4P release of CA4P NPs-3 and CA4P NPs-4.



**Fig. 4** (A) Tumor volume and (B) body weight changes during the treatment; (C) Tumor growth trend of each mouse; (D) Photos of the tumor bearing mice after various treatments. Results are presented as mean  $\pm$  SD; \* $p$ <0.05, \*\* $p$ <0.01, ns: not significant.



**Fig. 5** H&E staining of (A) tumors or (B) normal tissues after various treatments.

treated mice show rapidly proliferating cells, as indicated by distinct nuclei with abundant nucleoplasm and large nucleoli. Treatment with CA4P and CA4P NPs-4 significantly decreased tumor cell density, and partial necrotic areas can be observed in the CA4P group. No significant changes can be observed in normal tissues in any of the treatment groups (Fig. 5B), which further demonstrates that CA4P NPs-4 has excellent anti-tumor efficacy and biosafety.

## CONCLUSIONS

We designed multi-component CA4P polymeric nanoparticles via the salt bridge of phosphate group and guanidine group. The CA4P NPs-4 synthesized by the microfluidic method had high drug loading efficacy and stability, and achieved sustained release of CA4P, with translated to significantly higher tumor inhibition rate compared to CA4P, demonstrates a high safety profile and strong clinical translation potential.

## Conflict of Interests

The authors declare no interest conflict.

## Electronic Supplementary Information

Electronic supplementary information (ESI) is available free of charge in the online version of this article at <http://doi.org/10.1007/s10118-023-2921-7>.

## ACKNOWLEDGMENTS

This work was financially supported by the Ministry of Science and Technology of China (No. 2022YFE0110200), the Natural Science Foundation of Hunan Province of China (No. 2021JJ30680) and the National Natural Science Foundation of China (Nos. 52203198, 52025035 and 52103195).

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